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Schoonerite: its atomic arrangement

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Abstract

Schoonerite, $ZnMn^{2+}Fe_2^{2+}Fe^{3+}(OH)_2(H_2O)_7(PO_4)_3 \cdot 2H_2O$, orthorhombic, a 11.119(4), b 25.546(11), c 6.437(3) Å, $Pmab$, $Z = 4$, is one of several structurally-related phosphate minerals having axial repeats approximating 6.4 Å. The structure was solved by Patterson and Fourier syntheses, and least-squares refinement converged to $R(hkl) = 0.085$ for 897 nonequivalent reflections.

Edge-sharing Fe^{2+} -O octahedral chains join by sharing corners with Fe^{3+} -O octahedra and $[PO_4]$ tetrahedra forming a sheet parallel to {010}. To this sheet are linked additional $[PO_4]$ tetrahedra, $[MnO_6]$ octahedra and $[ZnO_5]$ polyhedra. The resultant slabs are joined in the b direction by a network of hydrogen bonds only. Interlayer water molecules contribute to this hydrogen bond network.

Average interatomic distances are $^6Fe^{2+}$ (1)-O 2.12 Å, $^6Fe^{2+}$ (2)-O 2.15, $^6Fe^{3+}$ (3)-O 2.00, $^6Mn^{2+}$ -O 2.16, 5Zn -O 2.09, $^4P(1)$ -O 1.54, and $^4P(2)$ -O 1.54.

Introduction

In the preceding paper, Moore and Kampf (1977) described the new species, schoonerite, $ZnMn^{2+}Fe_2^{2+}Fe^{3+}(OH)_2(H_2O)_7(PO_4)_3 \cdot 2H_2O$, $Z = 4$, from the Palermo No. 1 pegmatite in North Groton, New Hampshire. Although this species is rather widely distributed at Palermo, it occurs only sparingly as sprays of small, thin, feathery crystals in pockets and on fracture surfaces. A water determination was deemed unfeasible due to the difficulty in obtaining a sufficient quantity of pure material. The determination of the crystal structure of schoonerite was considered the best method for obtaining an accurate measure of its water content and for characterizing its rather complex crystal chemistry.

Besides accomplishing these ends, the crystal structure determination revealed a close similarity between the structures of schoonerite and montgomeryite, and in addition showed that schoonerite possesses octahedral and tetrahedral clusters locally similar to those in olmsteadite and melonjosephite.

Experimental

Good single crystals of schoonerite are exceedingly rare, and such crystals large enough for structure

analysis seemed for quite some time nonexistent. The crystal chosen, a thin rectangular tablet measuring $0.32 \times 0.01 \times 0.08$ mm along the three crystallographic axes, was mounted with the a axis parallel to the ϕ axis on a Picker automated diffractometer. The intensities of 2383 reflections (maximum $2\theta = 45^\circ$) were gathered, utilizing graphite-monochromatized $MoK\alpha$ radiation. A rather large mosaic spread necessitated wide half-angle scans of 1.0° with a scan rate of $2.0^\circ/\text{minute}$. Twenty-second background counting times were used on either side of each reflection. The least-squares refinement of 30 reference reflections ($2\theta = 20-30^\circ$) yielded cell constants, a 11.119(4), b 25.546(11), and c 6.437(3) Å. Systematic extinctions were consistent with either of the space groups $P2ab$ or $Pmab$.

The measured intensities were corrected for absorption using the Gaussian integral method described by Burnham (1966). The data were processed by conventional computational procedures. After averaging symmetry-equivalent reflections and rejecting all $I(hkl) < 3\sigma [I(hkl)]$, 897 independent data were available for the ensuing analysis.

Solution and refinement of the structure

The Patterson map, $P(uvw)$, revealed strong vector concentrations at the levels $u = 0, \frac{1}{4},$ and $\frac{1}{2}$, indicating that the twenty heavy atoms (Fe, Mn, Zn) in

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SCHCKNREFITE. CBSEFVEC AND CALCULATEE STRUCTURE FACTICES (XIE)

6

10	9	530	804	10	5	812	855	6	542	1035	4	3	1012	1060	3	5	581	653	4	1	511		
10	10	1662	1541	10	6	784	785	6	7	274	350	4	4	588	688	3	6	245	352	4	2	211	
10	11	700	675	10	8	500	444	6	6	304	260	4	5	362	349	3	7	1134	1221	5	3	515	
10	12	0	3426	10	9	638	444	6	6	536	696	4	6	354	285	3	8	1259	1264	5	3	501	
12	1	1259	1290	10	10	638	266	7	0	265	116	4	7	341	645	3	9	1259	1264	5	5	416	
12	2	1264	1160	11	1	320	455	7	1	893	875	4	5	435	471	3	10	1744	1681	5	5	461	
12	3	1744	1681	11	1	843	925	7	2	1767	1810	4	10	354	230	4	12	1346	1320	5	5	310	
12	4	1346	1320	11	2	752	808	7	3	564	606	5	0	496	434	4	12	1338	1328	5	5	252	
12	5	1338	1328	11	3	386	395	7	4	1366	1425	5	1	446	442	4	12	479	380	5	5	125	
12	6	479	380	11	4	364	385	7	5	1024	1023	5	2	255	182	4	12	609	941	5	5	1194	
12	7	609	941	11	5	323	323	7	6	1194	1224	5	4	464	368	4	12	1203	1150	5	5	348	
12	8	1203	1150	11	6	251	243	8	4	341	401	5	5	774	751	4	12	9	824	759	5	5	276
12	9	9	824	11	7	275	245	7	9	1353	1378	6	0	227	227	4	12	10	638	466	5	5	308
12	10	638	466	11	10	346	236	7	10	637	605	5	6	266	266	4	12	125	3185	5	5	208	
14	0	3185	3185	12	C	327	13C	7	11	258	394	5	8	278	56	5	14	1345	1292	5	5	14	
14	1	1345	1292	14	1	365	400	8	1	548	526	5	8	344	360	5	14	1204	1123	5	5	16	
14	2	1204	1123	12	2	251	243	8	4	1353	1378	6	0	1066	1066	4	14	4	1926	1925	5	5	1266
14	3	4	1926	1925	12	4	515	477	8	5	550	555	5	6	1241	1241	4	14	1768	1742	5	5	1266
14	4	1768	1742	12	6	372	255	8	7	401	465	6	2	1251	1251	4	14	875	755	5	5	1234	
14	5	875	755	13	C	311	230	8	6	374	480	6	3	776	56	5	14	14	419	276	5	5	1234
14	6	14	419	13	1	276	1064	8	10	1036	1036	5	4	250	140	6	14	1592	1522	5	5	1266	
14	7	1592	1522	13	2	703	727	5	6	1038	1066	6	4	1241	1241	4	14	1269	1151	5	5	1266	
14	8	1269	1151	13	3	434	458	9	1	1180	1218	6	6	1180	1180	4	14	1639	1523	5	5	1266	
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16	5	1392	1272	14	1	456	414	9	8	765	750	7	2	336	336	7	16	1613	1603	5	5	1266	
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18	7	480	456	16	7	616	616	10	10	380	341	8	7	513	553	8	18	441	406	5	5	1266	
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0	4	698	132	15	4	451	408	14	1	855	797	13	0	221	351	12	6	598	682	4	2	528	528
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0	10	1030	1023	15	7	722	682	14	5	760	756	13	6	323	425	15	2	842	546	5	6	582	582
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1	1	1030	1122	15	9	205	142	14	7	396	307	13	8	230	163	13	4	959	959	5	6	494	441
1	2	1030	1122	15	10	599	545	14	1	253	325	12	5	272	29	13	6	641	596	6	3	334	341
1	3	1030	1122	15	11	567	603	14	2	358	380	13	7	840	13	13	7	846	856	6	3	933	547
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1	10	1030	1122	15	18	956	356	16	4	956	1750	16	6	1025	1025	16	1	1841	1841	6	5	471	471
1	11	1030	1122	15	19	428	570	16	5	718	724	16	2	1950	1969	15	2	285	87	9	2	281	416
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1	13	1030	1122	15	21	501	574	17	0	857	849	16	5	518	508	16	4	1175	1145	10	6	967	923
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1	15	1030	1122	15	23	501	574	17	4	1543	1542	16	8	534	581	16	4	490	509	10	5	317	413
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